

## An Unnatural Amino Acid that Induces $\beta$ -Sheet Folding and Interaction in Peptides

James S. Nowick,<sup>\*,†</sup> Kit S. Lam,<sup>‡</sup> Tatyana V. Khasanova,<sup>†</sup> William E. Kemnitzer,<sup>†</sup> Santanu Maitra,<sup>†</sup> Hao T. Mee,<sup>†</sup> and Ruiwu Liu<sup>‡</sup>

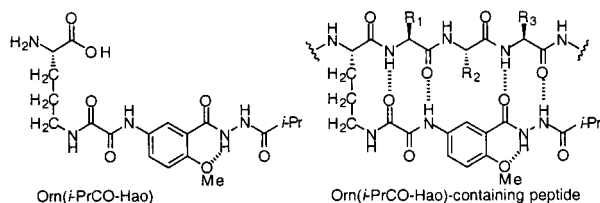
Department of Chemistry, University of California Irvine, Irvine, California 92697-2025, and Department of Internal Medicine, University of California Davis Cancer Center, Sacramento, California 95817

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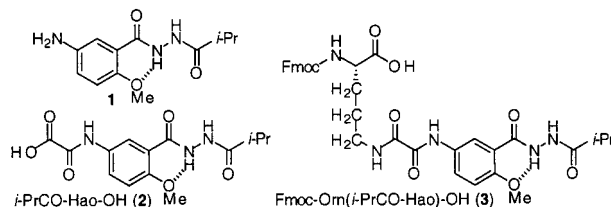
Peptides generally do not participate in the sort of strong and specific  $\beta$ -sheet interactions that occur widely among proteins.<sup>1</sup> While  $\beta$ -sheet formation between proteins is often involved in exquisite protein–protein interactions,  $\beta$ -sheet formation between peptides is generally associated with the formation of insoluble aggregates of ill-defined structure. Although a variety of elegant strategies to coax peptides to fold into well-defined  $\beta$ -sheetlike structures have now been developed, few strategies for achieving strong and specific  $\beta$ -sheet interactions between peptides have been reported.<sup>2,3</sup> Our (J.S.N.) research group has recently established that compounds that mimic the structures of protein  $\beta$ -sheets can dimerize in a highly specific fashion and has shown the facility with which amino acid building blocks that mimic the hydrogen-bonding functionality of a peptide  $\beta$ -strand can be used to generate peptides that dimerize by  $\beta$ -sheet formation.<sup>4,5</sup> Here, we introduce a unique amino acid that can readily be incorporated into peptides to make them fold into  $\beta$ -sheetlike structures that dimerize through  $\beta$ -sheet interactions.

This new amino acid, Orn(*i*-PrCO-Hao), consists of an ornithine residue with the  $\beta$ -strand-mimicking amino acid Hao<sup>5</sup> attached to its side chain. When Orn(*i*-PrCO-Hao) is incorporated into a peptide, or appended to its *N*-terminus, the Hao group hydrogen bonds to the three subsequent residues to form a  $\beta$ -sheetlike structure. The CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>NH side-chain of the ornithine residue allows the Hao oxalamide carbonyl group to form a hydrogen-bonded ten-membered ring with the amino group of the subsequent residue, like a  $\beta$ -turn in a  $\beta$ -hairpin. Chart 1 illustrates the structure of Orn(*i*-PrCO-Hao) and that of a peptide containing it.

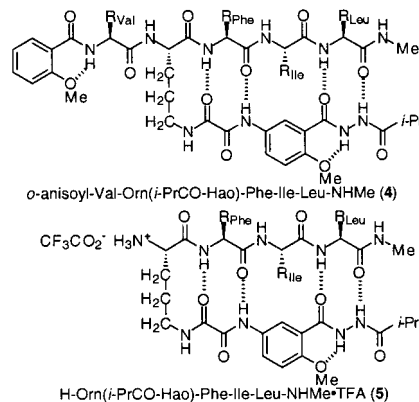
Chart 1



The amino acid Orn(*i*-PrCO-Hao) is readily used in peptide synthesis as its Fmoc derivative, which is conveniently prepared from hydrazide **1**.<sup>6</sup> Condensation of hydrazide **1** with ethyl oxalyl chloride, followed by hydrolysis of the ethyl ester group with NaOH and passage of the reaction mixture through acidic ion-exchange resin, yields *i*-PrCO-Hao-OH (**2**). Coupling of *i*-PrCO-Hao-OH and Boc-Orn-*O*-*t*-Bu with EDC and HOBt, followed by cleavage of the Boc and *tert*-butyl ester groups with TFA and Fmoc protection with Fmoc-OSu, yields Fmoc-Orn(*i*-PrCO-Hao)-OH (**3**).



Fmoc-Orn(*i*-PrCO-Hao)-OH behaves like a regular amino acid in peptide synthesis and was uneventfully incorporated into the peptide *o*-anisoyl-Val-Orn(*i*-PrCO-Hao)-Phe-Ile-Leu-NHMe (**4**) through standard automated Fmoc solid-phase peptide synthesis using PS-PEG-indole-NHMe<sup>7</sup> resin, with DIC and HOAt<sup>8</sup> as the coupling agent for Fmoc-Orn(*i*-PrCO-Hao)-OH and *o*-anisic acid.<sup>9</sup> A second synthetic strategy was developed to facilitate the preparation of peptides with *N*-terminal Orn(*i*-PrCO-Hao) residues, which avoids the need for the preparation of Fmoc-Orn(*i*-PrCO-Hao)-OH. In this strategy, Boc-Orn(Fmoc)-OH is used as the penultimate amino acid in the peptide synthesis, and *i*-PrCO-Hao-OH (**2**) is used as the final amino acid. *N*-Terminal Orn(*i*-PrCO-Hao) peptide H-Orn(*i*-PrCO-Hao)-Phe-Ile-Leu-NHMe·TFA (**5**) was prepared in a fashion similar to that of **4**, using DIC and HOAt<sup>8</sup> as the coupling agent for *i*-PrCO-Hao-OH.



<sup>1</sup>H NMR spectroscopic studies establish that peptide **4** forms a dimeric  $\beta$ -sheetlike structure in CDCl<sub>3</sub> solution. Most notably, <sup>1</sup>H NMR transverse-ROESY (Tr-ROESY)<sup>10</sup> studies in CDCl<sub>3</sub> solution (3.9 mM, 298 K) show five prominent NOEs between the *i*-PrCO-Hao unit and the Phe, Ile, Leu, and NHMe groups of the peptide strand indicative of a folded  $\beta$ -sheetlike structure and three prominent NOEs between the *N*- and *C*-terminal groups of the peptide (*o*-anisoyl-Val and Leu-NHMe) consistent with a dimeric structure. Chart 2 illustrates the structure of the dimer and shows these NOEs. Particularly compelling among these data is a strong

\* To whom correspondence should be addressed. E-mail: jsnowick@uci.edu.

<sup>†</sup> University of California, Irvine.

<sup>‡</sup> University of California Davis Cancer Center.

Chart 2

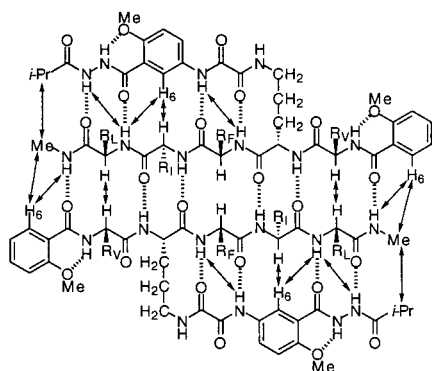
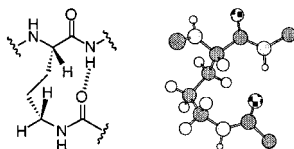


Chart 3



interstrand NOE between the proton at the 6-position of the aromatic group of Hao and the Ile  $\alpha$ -proton and a strong intersheet NOE between the Val and Leu  $\alpha$ -protons.

The  $^3J_{\text{HN}\alpha}$  coupling constants of the peptide strand of peptide **4** are large (8.8–9.8 Hz), and the interresidue NOEs between the NH and  $\alpha$ -protons of the amino acids are strong, reflecting a  $\beta$ -sheetlike conformation of the peptide strand.<sup>11–13</sup> Also consistent with  $\beta$ -sheetlike structure, the chemical shifts of the  $\alpha$ -protons of the amino acids in **4** are shifted 0.71–1.54 ppm downfield of values typical of random coil conformations.<sup>14</sup> Consistent with the hydrogen bonding associated with folding and dimerization in chloroform solution, the NH groups of the peptide strand appear at 8.3–9.4 ppm, which is substantially downfield of the characteristic chemical shift of peptide NH groups that are not hydrogen-bonded (ca. 6 ppm). Also consistent with a folded structure in which the Ile side-chain of the peptide is close to the aromatic ring of the Hao unit, the Ile  $\delta$ - and  $\gamma$ -methyl groups appear unusually upfield at 0.54 and 0.68 ppm, and one of the Ile  $\gamma$ -methylene protons appears unusually upfield at 0.75 ppm, while the other appears at 1.23 ppm.

The  $^1\text{H}$  NMR studies also suggest that the ornithine unit adopts a well-defined turn conformation. Particularly prominent in the Tr-ROESY spectrum of **4** is a *very strong* NOE between the  $\alpha$ -proton of Orn and one of its two diastereotopic  $\delta$ -protons. This proton appears dramatically downfield of the other diastereotopic  $\delta$ -proton (4.39 ppm vs 3.09), indicating that these two protons reside in very different environments. No NOE is seen between the  $\alpha$ -proton of Orn and the other diastereotopic  $\delta$ -proton, confirming the difference in environments. These two protons also exhibit widely different coupling patterns, with the downfield  $\delta$ -proton resonance resembling a quartet or a triplet of doublets with three large (10–12 Hz) coupling constants, and the upfield  $\delta$ -proton resembling a broad doublet with one large coupling constant. Molecular modeling gives a turn structure consistent with these NOE data and shift data, in which the *pro-S*  $\delta$ -proton is in contact with the Orn  $\alpha$ -proton and is shifted downfield by proximity to the adjacent carbonyl group of the Hao group and by proximity to the carbonyl of the adjacent Val residue. Chart 3 illustrates this turn structure.

Analogous  $^1\text{H}$  NMR studies of peptide **5** indicate that this TFA salt folds but does not dimerize in  $\text{CD}_3\text{OD}$  solution. Most notably, Tr-ROESY studies in  $\text{CD}_3\text{OD}$  solution (10.1 mM, 298 K) show an NOE between proton at the 6-position of the aromatic group of

Hao and the Ile  $\alpha$ -proton, indicating folding. NOEs between the Phe and Leu  $\alpha$ -protons, which would be expected in the antiparallel  $\beta$ -sheet dimer are absent. NOEs between the *i*-PrCO and MeN protons cannot be detected, due to overlap of the *i*-PrCO methine and MeN resonances, but are detected in 1:1  $\text{CD}_3\text{OD}$ - $\text{CDCl}_3$ , in which these resonances do not overlap. Consistent with proximity between the Ile side-chain and the aromatic ring of the Hao unit, the Ile  $\delta$ - and  $\gamma$ -methyl groups appear upfield at 0.70 and 0.74 ppm, respectively, and one of the Ile  $\gamma$ -methylene protons appears upfield at 0.85 ppm, while the other appears at 1.37 ppm. That these NOEs are weaker than those of **4** and the upfield shifting is more modest suggests that the extent of folding of **5** in  $\text{CD}_3\text{OD}$  is somewhat less than that of **4** in  $\text{CDCl}_3$ . Consistent with a less well-defined ornithine turn structure, the diastereotopic  $\delta$ -protons of **5** in  $\text{CD}_3\text{OD}$  are less separated than those of **4** in  $\text{CDCl}_3$ , appearing at 3.64 and 3.22 ppm.

Collectively, these synthetic and spectroscopic studies establish that the amino acid Orn(*i*-PrCO-Hao) induces  $\beta$ -sheet structure and interactions in peptides in suitable organic solvents. Unlike our Hao amino acid, which acts as a *prosthetic* to replace three residues of the peptide strand, the Orn(*i*-PrCO-Hao) amino acid acts as a *split* that helps enforce a  $\beta$ -sheetlike structure without replacing the residues and their side chains. This feature of Orn(*i*-PrCO-Hao) is important, because it allows the creation of  $\beta$ -sheet structure with minimal perturbation of the peptide sequence. Another important feature, which has allowed us to screen and analyze one-bead-one-compound combinatorial libraries of peptides containing Orn(*i*-PrCO-Hao) for biological activity, is that Orn(*i*-PrCO-Hao) behaves like a regular  $\alpha$ -amino acid in Edman sequencing.<sup>15</sup> We will describe this work in due course.

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**Supporting Information Available:** Synthetic procedures and spectral data (PDF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

## References

- (1) Maitra, S.; Nowick, J. S. In *The Amide Linkage: Structural Significance in Chemistry, Biochemistry, and Materials Science*; Greenberg, A., Breneman, C. M., Liebman, J. F., Eds.; Wiley: New York, 2000; Chapter 15.
- (2) Venkatraman, J.; Shankaramma, S. C.; Balaram, P. *Chem. Rev.* **2001**, *101*, 3131–3152.
- (3) Phillips, S. T.; Rezac, M.; Abel, Y.; Kossenjans, M.; Bartlett, P. A. *J. Am. Chem. Soc.* **2002**, *124*, 58–66.
- (4) Nowick, J. S.; Tsai, J. H.; Bui, Q.-C. D.; Maitra, S. *J. Am. Chem. Soc.* **1999**, *121*, 8409–8410.
- (5) Nowick, J. S.; Chung, D. M.; Maitra, K.; Maitra, S.; Stigers, K. D.; Sun, Y. *J. Am. Chem. Soc.* **2000**, *122*, 7654–7661.
- (6) (a) Nowick, J. S.; Pairish, M.; Lee, I. Q.; Holmes, D. L.; Ziller, J. W. *J. Am. Chem. Soc.* **1997**, *119*, 5413–5424. (b) Holmes, D. L.; Smith, E. M.; Nowick, J. S. *J. Am. Chem. Soc.* **1997**, *119*, 7665–7669.
- (7) Estep, K. G.; Neipp, C. E.; Stephens Stramiello, L. M.; Adam, M. D.; Allen, M. P.; Robinson, S.; Roskamp, E. *J. Org. Chem.* **1998**, *63*, 5300–5301.
- (8) Carpino, L. A. *J. Am. Chem. Soc.* **1993**, *115*, 4397–4398.
- (9) The *o*-anisoyl group enhances the solubility of peptide **4** and minimizes its aggregation by intramolecularly hydrogen bonding to the valine NH group.
- (10) (a) Hwang, T.-L.; Shaka, A. J. *J. Am. Chem. Soc.* **1992**, *114*, 3157–3159. (b) Hwang, T.-L.; Shaka, A. J. *J. Magn. Reson. Series B* **1993**, *102*, 155–165.
- (11) Wüthrich, K. *NMR of Proteins and Nucleic Acids*; Wiley: New York, 1986; pp 125–129.
- (12) Nowick, J. S.; Smith, E. M.; Pairish, M. *Chem. Soc. Rev.* **1996**, *25*, 401–415.
- (13) Wüthrich, K. *NMR of Proteins and Nucleic Acids*; Wiley: New York, 1986; pp 166–168.
- (14) (a) Wishart, D. S.; Sykes, B. D.; Richards, F. M. *J. Mol. Biol.* **1991**, *222*, 311–333. (b) Wishart, D. S.; Sykes, B. D.; Richards, F. M. *Biochemistry* **1992**, *31*, 1647–1651.
- (15) Lam, K. S.; Lebl, M.; Krchnák, V. *Chem. Rev.* **1999**, *97*, 4411–4448. JA025699I